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Carbon monoxide is the simplest oxocarbon and is isoelectronic with other triply bonded diatomic species possessing 10 valence electrons including the cyanide anion the nitrosonium cation boron monofluoride and molecular nitrogen. What is carbon monoxide CO and how is it produced? Carbon monoxide CO is a deadly colorless odorless poisonous gas.

It is produced by the incomplete burning of various fuels including coal wood charcoal oil kerosene propane and natural gas. Carbon monoxide CO is a colorless practically odorless and tasteless gas or liquid. It results from incomplete oxidation of carbon in combustion.

Burns with a violet flame. Slightly soluble in water soluble in alcohol and benzene. Carbon monoxide CO is a poisonous colorless odorless and tasteless gas. Although it has no detectable odor CO is often mixed with other gases that do have an odor.

So you can inhale carbon monoxide right along with gases that you can smell and not even know that CO is present. Carbon monoxide CO is an odorless colorless gas that kills without warning.

It claims the lives of hundreds of people every year and makes thousands more ill. Carbon monoxide CO is a highly toxic colorless odorless flammable gas produced industrially for use in the manufacture of numerous organic and inorganic chemical products. CO definition 1. together with 2.

Abbreviation for company = business 3. written abbreviation for county. Learn more.

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