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In chemistry an ester is a compound derived from an acid either organic or inorganic in which the hydrogen atom H of at least one acidic hydroxyl group OH of that acid is replaced by an organyl group R

[1] These compounds contain a distinctive functional group Dec 17 2025 Ester any of a class of organic compounds that react with water to produce alcohols and organic or inorganic acids. Esters derived from carboxylic acids are the most common

Learn about the different types and reactions of esters and more in this article Jun 9 2025 An ester is an organic compound where the hydrogen in the compound's carboxyl group is replaced with a hydrocarbon group. Esters are derived from carboxylic acids and usually alcohol Aug 23 2025 Esters are also the building blocks for an important class of polymers known as polyesters

These materials characterized by repeating ester linkages in their molecular chains are widely used in textiles such as polyethylene terephthalate PET fibers and in the production of plastic bottles Key Takeaway An ester has an OR group attached to the carbon atom of a carbonyl group Dec 1 2025 An ester is an organic compound that is a derivative of a carboxylic acid in which the hydrogen atom of the hydroxyl group has been replaced with an alkyl group Jul 23 2025 The general structure of an ester is RCOOR' where R and R' represent alkyl or aryl groups. Esters are derived from the condensation reaction between a carboxylic acid and an alcohol resulting in the elimination of water. We'll start with a very simple ester like ethyl ethanoate not something complicated like a fat or oil! The diagram shows the relationship between the ethanoic acid, the ethanol, and the ester. The ester linkage is also present in animal fats and in many biologically important molecules

The chemical industry uses esters for a variety of purposes. Ethyl acetate for instance is a commonly used solvent and dialkyl phthalates are used as plasticizers to keep polymers from becoming brittle. In an ester, the second oxygen atom bonds to another carbon atom (Figure 25.5a)

The names for esters include prefixes that denote the lengths of the carbon chains in the molecules and are derived following nomenclature rules similar to those for inorganic acids and salts.

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