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Silicon is the eighth most common element in the universe by mass but very rarely occurs in its pure form in the Earth's crust

It is widely distributed throughout space in cosmic dusts, planetoids and planets as various forms of silicon dioxide, silica or silicates. Silicon is a nonmetallic chemical element in the carbon family that makes up 27.7 percent of Earth's crust. It is the second most abundant element in the crust, being surpassed only by oxygen.

Learn more about the characteristics, distribution and uses of silicon in this article. Element Silicon Si Group 14, Atomic Number 14, p block, Mass 28.085. Sources, facts, uses, scarcity, SRI, podcasts, alchemical symbols, videos and images. Silicon is a brittle and hard, crystalline solid.

It has a blue-grey metallic lustre. Silicon, in comparison with neighbouring elements in the periodic table, is unreactive. The symbol for silicon is Si with atomic number 14.

It has a very high melting and boiling point. Silicon is prepared commercially by heating silica and carbon in an electric furnace using carbon electrodes.

Several other methods can be used for preparing the element. Silicon is the eighth most abundant element in the Universe; it is made in stars with a mass of eight or more Earth suns.

Near the end of their lives, these stars enter the carbon-burning phase, adding helium nuclei to carbon to produce oxygen, neon, magnesium and silicon. Silicon is the eighth most common element in the universe by mass but very rarely occurs in its pure form in the Earth's crust.

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